

Ex: For graphite,  $\rho = 2,3 \text{ g/cm}^3$ ,  $A = 12 \text{ g/mol}$ . Find the number of atoms per  $\text{cm}^3$  reminder:  $1 \text{ mole} = 6,023 \times 10^{23}$

$$n = \frac{N_A \times \rho}{A} = \frac{6,023 \times 10^{23} \times 2,3 \text{ g/cm}^3}{12 \text{ g/mol}} = 11,5 \times 10^{22} \text{ atom/cm}^3 ?$$

Ex: Calculate the number of atoms in 100g of Ag. (Reminder:  $A_{\text{Ag}} = 107,868 \text{ g/mol}$ )

$$\text{Number of in 100g Ag} = \frac{100 \cdot (6,023 \cdot 10^{23} \text{ atoms})}{107,868 \text{ g/mol}} = \underline{\underline{5,58 \cdot 10^{23} \text{ atoms}}}$$

Ex: Consider an alloy of 75 wt% Cu and 25 wt% Ni. What are the atomic percentage of Cu and the atomic percentage of Ni of this material?

(For Cu,  $A_{\text{Cu}} = 63,54 \text{ g/mol}$ )  
 (For Ni,  $A_{\text{Ni}} = 58,69 \text{ g/mol}$ )

75 % wt Cu      25 % wt Ni

100g      75g Cu      25g Ni

$$\frac{75 \text{ g}}{63,54 \text{ g/mol}} = 1,1803 \text{ mol Cu} \quad \frac{25 \text{ g}}{58,69 \text{ g/mol}} = 0,426 \text{ mol Ni} \quad \left. \vphantom{\frac{75 \text{ g}}{63,54 \text{ g/mol}}} \right\} \text{ Total} = 1,606 \text{ mol}$$

100 %      1,606 mol

100 %      1,606 mol

X %      1,1803 mol

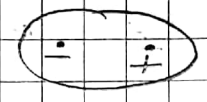
X %      0,426 mol

Cu = 73,5 at %

Ni = 26,5 at %

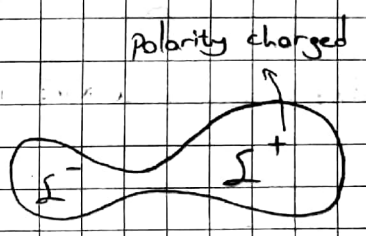
Ex:  $\overset{F}{4.0}$   $\overset{F}{4.0}$  Find the EN difference

$4.0 - 4.0 = 0$   $0.0 \sim 0.4 \Rightarrow$  Non-polar covalent



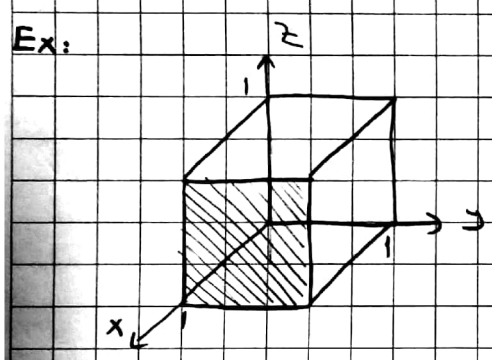
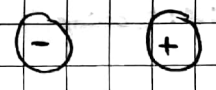
Ex:  $\overset{O}{3.5}$   $\overset{F}{4.0}$  Find the EN difference

$4.0 - 3.5 = 0.5$   $0.4 \sim 2.0$  then it is polar covalent



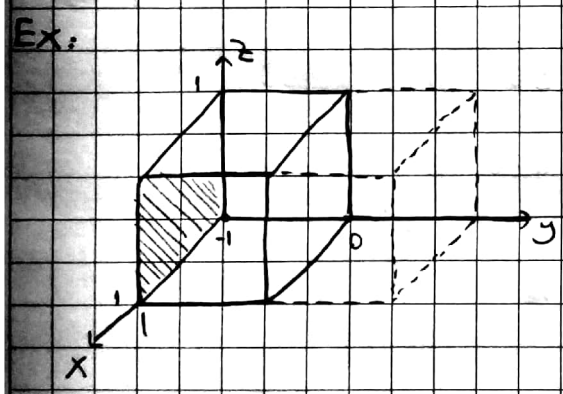
Ex:  $\overset{Ca}{1.0}$   $\overset{O}{3.5}$  Find the EN difference

$3.5 - 1.0 = 2.5 > 2.0$  then it is ionic bond



	x	y	z
intercepts	1	$\infty$	$\infty$
reciprocals	$\frac{1}{1}$	$\frac{1}{\infty}$	$\frac{1}{\infty}$
reduce the fraction	1	0	0

Miller indices (1 0 0)



x	y	z
$\infty$	-1	$\infty$
$\frac{1}{\infty}$	$\frac{1}{-1}$	$\frac{1}{\infty}$
0	-1	0

Miller indices (0  $\bar{1}$  0)  $\bar{1}$  - always in

Ex: A current density of  $100,000 \text{ A/cm}^2$  is applied to a gold wire  $50 \text{ m}$  long. The resistance of the wire found to be  $2 \Omega$ . Calculate the voltage applied to the wire, and calculate the diameter of the wire.

$$\left( \begin{array}{l} \rho_{Au} = 4,26 \cdot 10^5 \Omega^{-1} \text{ m}^{-1} \\ \pi = 3,14 \end{array} \right)$$

$$J = \frac{I}{A} = \sigma \cdot \frac{V}{l} \quad E = \frac{V}{l} \quad J = \sigma \cdot E$$

$$100000 = 4,26 \cdot 10^5 \cdot \frac{V}{50} \quad \boxed{V = 11,74 \text{ V}}$$

$$I = \frac{V}{R} = \frac{11,74}{2} = 5,87 \text{ A} \quad A = \frac{I}{J} = \frac{5,87}{100000} = 5,87 \cdot 10^{-5} \text{ cm}^2$$

$$A = \pi r^2 \Rightarrow r = 0,00432 \text{ m}$$

$$d = 2r = 8,64 \cdot 10^{-3} \text{ cm}$$

Ex: Choose a suitable material from the table below for an electrical transmission line of 1500m long that will carry a current of 50A with no more than  $5 \cdot 10^5$  W power loss.

	$\sigma$ ( $\Omega^{-1} \cdot m^{-1}$ )	Density ( $Mg \cdot m^{-3}$ )	Price (TL/tonne)	(1 Mg = 1 tonne)
Aluminium	$3,77 \cdot 10^7$	2,7	600	
Copper	$5,98 \cdot 10^7$	8,93	8500	
Silver	$6,8 \cdot 10^7$	10,49	480000	

$$P = I \cdot V = I^2 \cdot R \Rightarrow R = \frac{P}{I^2} = \frac{5 \cdot 10^5}{50^2} = 200 \Omega \quad R = \rho \cdot \frac{L}{A} \quad \rho = \frac{R \cdot A}{L}$$

$$A = \frac{L \cdot \rho}{R} = \frac{1500 \cdot \rho}{200} = \frac{7,5}{\rho}$$

$$A_{Al} = \frac{7,5}{3,77 \cdot 10^7} = 0,199 \text{ m}^2 \quad A_{Cu} = \frac{7,5}{5,98 \cdot 10^7} = 0,125 \text{ m}^2 \quad A_{Ag} = \frac{7,5}{6,8 \cdot 10^7} = 0,11 \text{ m}^2$$

$$\text{Weight of Al} = A \cdot L \cdot d = 8,06 \cdot 10^{-4} \text{ tonnes}$$

$$\text{cost Al} = 8,06 \cdot 10^{-4} \cdot 6000 = 4,83 \text{ TL}$$

$$\text{Weight of Cu} = A \cdot L \cdot d = 1,67 \cdot 10^{-3} \text{ tonnes}$$

$$\text{cost Cu} = 14,185 \text{ TL}$$

$$\text{Weight of Ag} = A \cdot L \cdot d = 1,73 \cdot 10^{-3} \text{ tonnes}$$

$$\text{cost Ag} = 847,7 \text{ TL}$$

Based on this analysis, Al is the most economical choice even though it has large diameter.

EX: Calculate the electrical conductivity of pure copper at

a) 400 °C

b) -100 °C

$\rho$ : resistivity at a temperature (T)

reminders: Resistivity of copper at 25 °C  $1,67 \cdot 10^{-8} \Omega \cdot m$

The temperature resistivity coefficient  $\alpha = 0,0068 \text{ } ^\circ\text{C}^{-1}$

At 400 °C

$$\rho = \rho_0 [1 + \alpha (T - T_0)] \quad \rho = (1,67 \cdot 10^{-8}) [1 + 0,0068 (400 - 25)]$$

$$\rho = 5,829 \cdot 10^{-8} \Omega$$

$$\sigma = \frac{1}{\rho} = \underline{1,69 \cdot 10^7 \text{ } \Omega^{-1} \cdot m^{-1}}$$

At -100 °C

$$\rho = (1,67 \cdot 10^{-8}) [1 + 0,0068 (-100 - 25)] \quad \rho = 0,251 \cdot 10^{-8} \Omega \quad \sigma = \underline{39,8 \cdot 10^7 \text{ } \Omega^{-1} \cdot m^{-1}}$$

Ex: For germanium at 25 °C estimate

Reminders:  $\sigma = 2,0 \text{ } \Omega^{-1} \cdot m^{-1}$

a) the number of charge carriers

$$E_g = 0,67 \text{ eV}$$

b) the constant,  $n_0$

Mobility  $\leftarrow \mu_e = 0,38 \text{ } m^2 V^{-1} s^{-1}$

$$\mu_h = 0,18 \text{ } m^2 V^{-1} s^{-1}$$

$$2kT = 0,0514 \text{ eV at } 25^\circ\text{C}$$

Boltzmann Constant  $\leftarrow$

$$q = 1,6 \cdot 10^{-19} \text{ } C$$

a)  $\sigma = n \cdot q (\mu_e + \mu_h) \quad n = \frac{2,0}{(1,6 \cdot 10^{-19})(0,38 + 0,18)} = 2,2 \cdot 10^{19} \text{ carriers } / m^3$

b)  $n = n_0 \cdot \exp\left(-\frac{E_g}{2kT}\right) \quad n_0 = \frac{2,2 \cdot 10^{19}}{\exp(0,67/0,0514)} = 1,01 \cdot 10^{25} \text{ carriers } / m^3$

e  $\leftarrow$



Ex: For Si, if the conductivity is  $5 \cdot 10^{-6} \Omega^{-1} \text{cm}^{-1}$  at  $25^\circ\text{C}$ . Calculate the temperature required to double the electrical conductivity.

Reminders:  $\mu_e = 1300 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$

$\mu_h = 500 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$

$n_0 = 2,835 \cdot 10^{19} \text{ carriers/cm}^3$

$k = 8,63 \cdot 10^{-5} \text{ e.V. K}^{-1}$  (change to  $^\circ\text{C}$ )

$q = 1,6 \cdot 10^{-19} \text{ C}$

$E_g = 1,107 \text{ eV}$

$\sigma = 5 \cdot 10^{-6} \Omega^{-1} \text{cm}^{-1}$  at  $25^\circ\text{C}$

$$\sigma = n \cdot q (\mu_e + \mu_h)$$

$$2 \cdot (5 \cdot 10^{-6}) = n \cdot (1,6 \cdot 10^{-19}) (1300 + 500) \Rightarrow n = 2,604 \cdot 10^{10}$$

$$n = n_0 \cdot \exp\left(-\frac{E_g}{2kT}\right) \quad 2,604 \cdot 10^{10} = 2,835 \cdot 10^{19} \cdot \exp\left(-\frac{1,107}{2 \cdot 8,63 \cdot 10^{-5} \cdot T}\right)$$

$$\ln(8,834 \cdot 10^{-10}) = \left(\exp\left(-\frac{1,107}{1,726 \cdot 10^{-4} T}\right)\right) \ln \Rightarrow -20,5228 = -\frac{1,107}{1,726 \cdot 10^{-4} T}$$

$$T = 307,92 \text{ K} = 307,92 - 273 = \underline{\underline{34,92^\circ\text{C}}}$$

Ex: In order to keep the resistance of a copper wire to  $0,5 \Omega$  where the length is  $30 \text{ m}$ . How long the radius of the wire should be?

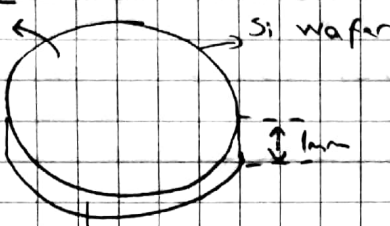
( $\rho = 1,68 \cdot 10^{-8} \Omega \text{m}$ )

$$R = \rho \frac{L}{A} \quad A = \rho \cdot \frac{L}{R} \quad A = 1,008 \cdot 10^{-6} \text{ m}^2 \quad A = \pi r^2 \Rightarrow \underline{\underline{r = 5,66 \cdot 10^{-4} \text{ m}}}$$

Ex: Suppose a silicon wafer, 1 mm thick, which originally contains 1 phosphorus atom for every 10 million silicon atoms, is treated so that there are 400 phosphorus atoms for every 10 million silicon atoms at the surface.

Calculate the concentration gradient in atomic percent.  $m^{-1}$ .

$C_{\text{surface}} = \frac{400P}{10^7 Si}$   
 ↓  
 After doping of phosphorus



Si wafer  
 1 mm  
 $C_{\text{initial}} = \frac{1P}{10^7 Si}$   
 $C_{\text{surface}} = \frac{400P}{10^7 Si}$

Concentration gradient =  $\frac{\Delta C}{\Delta x}$   
 $C_i = \frac{1}{10^7} \cdot 100 = 0,0001 \text{ at \% P}$   
 $C_{\text{surface}} = \frac{400}{10^7} \cdot 100 = 0,004 \text{ at \% P}$

$$\text{Concentration gradient} = \frac{(0,0001 - 0,004) \text{ at \% P}}{0,001 \text{ m}} = \underline{\underline{-3,99 \text{ at \% P} \cdot m^{-1}}}$$

Ex: When a Cu-Zn alloy solidifies, one portion of structure contains 25 atomic percent Zinc and another portion, 0,025 mm away, 20 atomic percent Zinc. Determine the concentration gradient.

- a) In atomic percent Zn per cm. (Reminder: Atomic mass of Zn = 65,38 g/mol)  
 b) In weight percent Zn per cm. (Atomic mass of Cu = 63,54 g/mol)

$$\text{a) } \frac{\Delta c}{\Delta x} = \frac{20 \text{ at \%} - 25 \text{ at \%}}{(0,025 \text{ cm})} = -2000 \text{ at \% Zn} \cdot \text{cm}^{-1}$$

b) We need to define the wt % of each portion.

$$\text{wt \% Zn}_1 = \frac{20 \cdot 65,38}{20 \cdot (65,38) + 80 \cdot (63,54)} = 20,46 \text{ wt \% Zn}$$

$$\text{wt \% Zn}_2 = \frac{25 \cdot 65,38}{25 \cdot (65,38) + 75 \cdot (63,54)} = 25,54 \text{ wt \% Zn}$$

$$\frac{\Delta c}{\Delta x} = \frac{20,46 \% - 25,54 \%}{0,025 \text{ cm}} = -2032 \text{ wt \% Zn} \cdot \text{cm}^{-1}$$

Ex: Consider a diffusion couple, set up between pure tungsten and a tungsten - 1 at% thorium alloy. After several minutes of exposure at  $2000^{\circ}\text{C}$ , a transition zone of  $0,1 \text{ m}$  thickness is established, what is the flux of thorium atoms at this time if  $D_0 = 1,00 \times 10^{-4}$ ,  $Q = 502300 \text{ J/mol}$  for thorium, in tungsten. And the number of tungsten atoms is  $6,3 \cdot 10^{28} \text{ atoms m}^{-3}$ ,  $R = 8,314 \text{ J/mol K}$  the number of thorium atoms,  $C_{\text{thorium}} = (0,01) \cdot (6,3 \cdot 10^{28}) = 6,3 \cdot 10^{26} \text{ atoms m}^{-3}$  in pure tungsten the number thorium atoms is zero.

$$\text{Thus } \frac{\Delta c}{\Delta x} = \frac{0 - 6,3 \cdot 10^{26}}{0,1 \cdot 10^{-3}} = -6,3 \cdot 10^{30} \text{ atoms m}^{-3} \cdot \text{m}^{-1}$$

$$D = D_0 \cdot \exp\left(-\frac{Q}{RT}\right) = 1,0 \cdot 10^{-4} \cdot \exp\left(-\frac{502300}{8,314 \cdot 2273}\right) \Rightarrow D = 2,86 \cdot 10^{-16} \text{ m}^2 \cdot \text{s}^{-1}$$

↳ Kelvin  $C + 273$

$$J = -D \cdot \frac{\Delta c}{\Delta x} \quad J = 18,0 \cdot 10^{14} \text{ Th atoms m}^{-2} \cdot \text{s}^{-1}$$

0,014

Ex: A parallel plate capacitor by two metal plates, each area of  $140 \text{ cm}^2$ , separated by vacuum, and gap between plates is  $d = 0,6 \text{ cm}$ . What is the capacitance of this device? What potential difference must be applied between plates if the capacitor is to hold a charge of  $Q = 0,8 \cdot 10^{-3} \text{ } \frac{\mu\text{C}}{10^{-6}}$  on each plate? ( $\epsilon_0 = 8,85 \cdot 10^{-12} \text{ F/m}$ )

$$C_0 = \frac{\epsilon_0 \cdot A}{d} = \frac{8,85 \cdot 10^{-12} \cdot 0,014}{0,006} = \underline{\underline{20,65 \text{ pF}}}$$

$$D_0 = \frac{Q_0}{A} = 5,714 \cdot 10^{-7} \rightarrow D_0 = \frac{C_0 \cdot V}{A} \Rightarrow \underline{\underline{V = 38,74 \text{ V}}}$$



Ex: Consider two insulator material, tested with breakdown experiment. Results is as following;

1. material breakdown 20 kV when the distance of plates was  $d_1 = 0,2 \text{ cm} = 0,002 \text{ m}$

2. material breakdown 250 kV when the distance of plates was  $d_2 = 0,1 \text{ cm} = 0,001 \text{ m}$

Which material you would choose for your high voltage application?

$$E_{d1} = \frac{V_{d1}}{d_1} = \frac{20}{0,2} = 100 \text{ kV/cm}$$

$$E_{d2} = \frac{V_{d2}}{d_2} = \frac{250}{0,1} = 2500 \text{ kV/cm}$$

}  $E_{d2} > E_{d1}$  so the 2. material would be better to choose.

Ex: A condenser system has  $C = 100 \text{ pF}$   $V = 100 \text{ V}$  with  $f = 1 \text{ MHz}$  and active loss power of  $P_a = 10^{-3} \text{ W}$ . Find the reactive power and loss tangent. ( $\pi = 3,14$ )

$$P_a = \underbrace{V^2 \cdot 2\pi f C \cdot \tan \delta}_{\text{reactive power (Pr)}} \quad \tan \delta = \frac{P_a}{P_r} \quad X_c = \frac{1}{2\pi f C}$$

$$10^{-3} = 100^2 \cdot 2\pi f C \cdot \tan \delta \quad P_r = 100^2 \cdot 2\pi f C = 6,28 \text{ VAR}$$

$$\tan \delta = \frac{10^{-3}}{100^2 \cdot 2 \cdot 3,14 \cdot 1 \cdot 10^6 \cdot 100 \cdot 10^{-12}} \quad \tan \delta = 1,6 \cdot 10^{-16}$$

Ex: A parallel plate capacitor with the surface area of  $2 \times 1,5 \text{ m}^2$  with separated distance at  $3 \text{ mm}$ . The gap between plates is filled with an insulator ( $\epsilon_r = 2$ ).

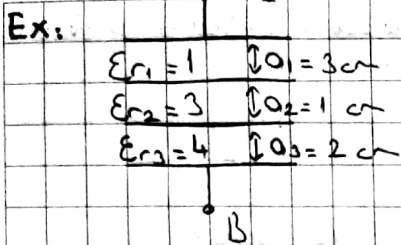
When the capacitor is fully charged with 12V battery. What will be the charge on plates? ( $\epsilon_0 = 8,85 \cdot 10^{-12}$ )

$$D = \frac{Q}{A} \quad D = \epsilon_0 \cdot \epsilon_r \cdot \frac{V}{d} \quad D = 8,85 \cdot 10^{-12} \cdot 2 \cdot \frac{12}{0,003} = 7,08 \cdot 10^{-8}$$

$$Q = D \cdot A \Rightarrow Q = 1,062 \cdot 10^{-7}$$

?

A  $\leftarrow V = 100 \text{ kV}$



Consider the system in figure

- Find the potential and electrical fields on each layer.
- Determine if the system would breakdown totally.

$$E_{d1} = 30 \text{ kV/cm}$$

$$E_{d2} = 30 \text{ kV/cm}$$

$$E_{d3} = 50 \text{ kV/cm}$$

$$A_1 = A_2 = A_3$$

$$K = \frac{q_1}{E_1} + \frac{q_2}{E_2} + \frac{q_3}{E_3} = \frac{q_1}{\epsilon_0 \cdot \epsilon_{r1}} + \frac{q_2}{\epsilon_0 \cdot \epsilon_{r2}} + \frac{q_3}{\epsilon_0 \cdot \epsilon_{r3}}$$

$$K = \left( \frac{q_1}{\epsilon_{r1}} + \frac{q_2}{\epsilon_{r2}} + \frac{q_3}{\epsilon_{r3}} \right) \frac{1}{\epsilon_0} \Rightarrow K = K_r \frac{1}{\epsilon_0} \quad K_r = \frac{3}{1} + \frac{1}{3} + \frac{2}{4} = 3,833$$

$$V_1 = \frac{q_1}{E_1} \cdot \frac{V}{K} = \frac{q_1}{\epsilon_0 \cdot \epsilon_{r1}} \cdot \frac{V}{K_r \frac{1}{\epsilon_0}} = \frac{q_1}{\epsilon_{r1}} \cdot \frac{V}{K_r} = \frac{3}{1} \frac{100}{3,833} = 26,267 \text{ kV} \quad E_1 = \frac{V_1}{d_1} = 26,083 \text{ kV/cm}$$

$$V_2 = \frac{q_2}{\epsilon_{r2}} \cdot \frac{V}{K_r} = 8,636 \text{ kV}$$

$$E_2 = \frac{V_2}{d_2} = 8,36 \text{ kV/cm}$$

$$V_3 = \frac{q_3}{\epsilon_{r3}} \cdot \frac{V}{K_r} = 13,044 \text{ kV}$$

$$E_3 = \frac{V_3}{d_3} = 6,522 \text{ kV/cm}$$

$$E_1 < E_{d1}$$

$$E_2 < E_{d2}$$

$$E_3 < E_{d3}$$

The system would not breakdown.

Ex: Consider a seriesly structured 4-layered parallel plate electrode system. Dielectric constant and thickness of the layers;

$$\epsilon_{r1} = 2 \quad \epsilon_{r2} = 1 \quad \epsilon_{r3} = 3 \quad \epsilon_{r4} = 4$$

applied voltage is  $V = 200 \text{ kV}$

$$a_1 = 1 \text{ cm} \quad a_2 = 2 \text{ cm} \quad a_3 = 1 \text{ cm} \quad a_4 = 4 \text{ cm}$$

a) Find the electrical fields and potentials on each layer.

b)  $E_{d1} = 100 \text{ kV/cm}$  find out if the system would breakdown totally?

$$E_{d2} = 30 \text{ kV/cm}$$

$$E_{d3} = 80 \text{ kV/cm}$$

$$E_{d4} = 60 \text{ kV/cm}$$

$$a) K = \frac{a_1}{\epsilon_1} + \frac{a_2}{\epsilon_2} + \frac{a_3}{\epsilon_3} + \frac{a_4}{\epsilon_4} = \underbrace{\left( \frac{a_1}{\epsilon_{r1}} + \frac{a_2}{\epsilon_{r2}} + \frac{a_3}{\epsilon_{r3}} + \frac{a_4}{\epsilon_{r4}} \right)}_{K_r} \frac{1}{\epsilon_0} \Rightarrow K = K_r \frac{1}{\epsilon_0} \quad K_r = 3,833$$

$$E_1 = \frac{V}{\epsilon_1 \cdot K} = \frac{V}{\epsilon_{r1} \cdot K_r \cdot \frac{1}{\epsilon_0}} = \frac{V}{\epsilon_{r1} \cdot K_r} = \frac{200}{2 \cdot 3,833} = 26,089 \text{ kV/cm}$$

$$E_2 = \frac{V}{\epsilon_{r2} \cdot K_r} = 52,178 \text{ kV/cm} \quad E_3 = \frac{V}{\epsilon_{r3} \cdot K_r} = 17,39 \text{ kV/cm} \quad E_4 = \frac{V}{\epsilon_{r4} \cdot K_r} = 13,044 \text{ kV/cm}$$

$$E_1 < E_{d1} \checkmark \quad E_2 > E_{d2} \Rightarrow 2. \text{ layer would breakdown} \quad E_3 < E_{d3} \checkmark \quad E_4 < E_{d4} \checkmark$$

b)  $E_2 > E_{d2}$  thus the second layer would breakdown and the total voltage would divide up to only 3 layers. In this case, we need to re-calculate for 3 layers.:

$$K_r = \frac{a_1}{\epsilon_{r1}} + \frac{a_2}{\epsilon_{r2}} + \frac{a_3}{\epsilon_{r3}} = 1,833$$

$$E_1 = \frac{V}{\epsilon_{r1} \cdot K_r} = 54,55 \text{ kV/cm}$$

$$E_1 < E_{d1}$$

$$E_3 < E_{d3}$$

$$E_4 < E_{d4}$$

Thus, the system would not breakdown totally. However, due to the breakdown of second layer, the system is weakened.

$$E_3 = \frac{V}{\epsilon_{r3} \cdot K_r} = 36,37 \text{ kV/cm}$$

$$E_4 = \frac{V}{\epsilon_{r4} \cdot K_r} = 27,28 \text{ kV/cm}$$

Ex: An Fe-80% Ni alloys has a maximum relative permeability of 300000 when an inductance of 3500 Gauss is obtained. The alloys is placed in a 20-turn coil that is 2cm in length. What current must flow through the conductor to obtain this field?

$$\left( \begin{array}{l} 1 \text{ Tesla} = 10^4 \text{ Gauss} \\ \mu_0 = 4\pi \times 10^{-7} \text{ Wb/A}\cdot\text{m} \end{array} \right) \quad \mu_r = \frac{\mu}{\mu_0} \Rightarrow 300000 = \frac{\mu}{4\pi \times 10^{-7}} \quad \mu = 0,3768 \text{ Wb/A}\cdot\text{m}$$

$$H = \frac{B}{\mu} = \frac{3500/10}{0,3768} = 0,928 \text{ A/m}$$

$$H = \frac{n \cdot I}{l} \Rightarrow I = \frac{H \cdot l}{n} = \frac{(0,928 \text{ A/m})(0,02 \text{ m})}{20} \Rightarrow I = 0,00093 \text{ A}$$

Ex: To produce a solenoid coil that has an inductance of 2000 Gauss when a 10mA current flows through the conductor. The coil composed of 10-turns over a 1cm length.

What would be the minimum relative permeability of material?

$$1 \text{ Tesla} = 10^4 \text{ Gauss}$$

$$\mu_0 = 4\pi \times 10^{-7} \text{ Wb/A}\cdot\text{m}$$

$$H = \frac{n \cdot I}{l} = \frac{10 \cdot 0,01 \text{ A}}{0,01 \text{ m}} = 10 \text{ A/m}$$

$$\mu = \frac{B}{H} = \frac{2000/10^4}{10} = 0,02 \text{ Wb/A}\cdot\text{m}$$

$$\mu_r = \frac{\mu}{\mu_0} = \frac{0,02}{4\pi \times 10^{-7}} = 15823$$

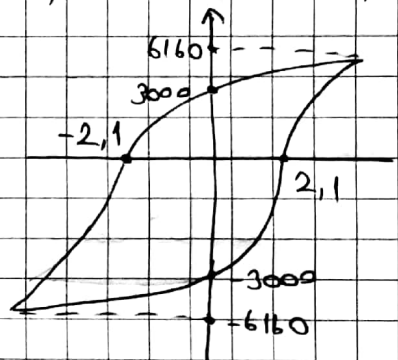
Ex: An Fe-43% Ni alloy has a maximum relative permeability of 6000 when a magnetic field of 3,347 A/m is applied. What inductance is obtained and what current is needed to obtain this inductance in 200-turn, 3cm long coil?

Reminder:  $\mu = 4\pi \cdot 10^{-7} \text{ Wb/A}\cdot\text{m}$

$$\mu_r = \frac{\mu}{\mu_0} \Rightarrow \mu = 0,0804 \frac{\text{Wb}}{\text{A}\cdot\text{m}} \quad B = 7,337 \text{ Tesla} \quad I = 1,43 \text{ A}$$

Ex: A magnetic material has a coercive field of 167 A/m, a saturation magnetization of 0,616 Tesla and a residual inductance of 0,3 Tesla. Sketch the hysteresis loop for the material in units of Gauss for B, and in Oersted for H?

Reminder  $1 \text{ Tesla} = 10^4 \text{ Gauss}$   
 $1 \text{ A/m} = 4\pi \cdot 10^{-3} \text{ Oe}$



$$B_{\text{saturation}} = 0,616 \text{ Tesla} = 6160 \text{ Gauss}$$

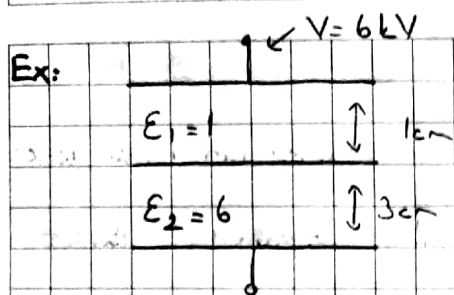
$$B_r = 0,3 \text{ Tesla} = 3000 \text{ Gauss} \quad H_c = 167 \text{ A/m} \cdot 4\pi \cdot 10^{-3} \text{ Oe/A}\cdot\text{m}^{-1}$$

$$H_c = 2,1 \text{ Oe}$$

Ex: Calculate the force in kN for one square meter area of a permanent magnet with a saturation magnetization of 1,61 Tesla

$$F = \frac{\mu_0 \cdot M^2 \cdot A}{2} = \frac{(\mu_0 \cdot M)^2 \cdot A}{2 \mu_0} = \frac{(1,61)^2 \cdot A}{2(4\pi \cdot 10^{-7})} \quad \frac{F}{A} = 1031,4 \frac{\text{kN}}{\text{m}^2}$$

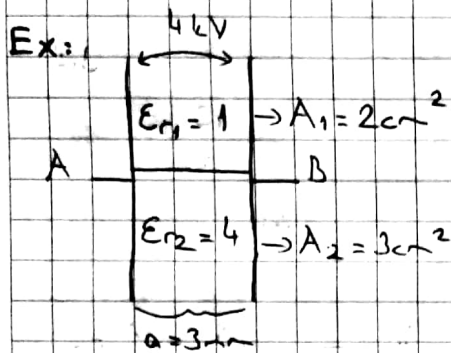




find the electric field on each insulator.  $E_1 = ?$   $E_2 = ?$

$$E_1 = \frac{V}{a_1 + \frac{E_1 \cdot a_2}{E_2}} = \frac{6000}{1 + \frac{1 \cdot 3}{6}} = 4000 \text{ V/cm}$$

$$E_2 = \frac{V}{a_2 + \frac{E_2 \cdot a_1}{E_1}} = \frac{6000}{3 + \frac{6 \cdot 1}{1}} = 666,66 \text{ V/cm}$$



A parallel structured condenser system with two dielectrics is given figure

a) Find the capacitance of each layer and equivalent capacitance in pF.

b) Find the electrical field on each layer and find the surface charge density.

$$(\epsilon_0 = 8,85 \cdot 10^{-12} \text{ F/m})$$

$$a) C_1 = \frac{\epsilon_1 \cdot A_1}{a_1} = \frac{\epsilon_0 \cdot \epsilon_{r1} \cdot A_1}{a_1} = \frac{8,85 \cdot 10^{-12} \cdot 1 \cdot 2 \cdot 10^{-4}}{3 \cdot 10^{-3}} = 5,9 \cdot 10^{-13} = 0,59 \text{ pF}$$

$$C_2 = \frac{\epsilon_2 \cdot A_2}{a_2} = \frac{\epsilon_0 \cdot \epsilon_{r2} \cdot A_2}{a_2} = \frac{8,85 \cdot 10^{-12} \cdot 4 \cdot 3 \cdot 10^{-4}}{3 \cdot 10^{-3}} = 3,54 \text{ pF}$$

$$C_{eq} = C_1 + C_2 = 0,59 + 3,54 = 4,13 \text{ pF}$$

$$b) E_1 = E_2 = \frac{V}{a} = \frac{4 \text{ kV}}{0,3 \text{ cm}} = 13,33 \text{ kV/cm} = 13,33 \cdot 10^5 \text{ V/m}$$

$$D_1 = \epsilon_1 \cdot E_1 = \epsilon_0 \cdot \epsilon_{r1} \cdot E_1 = 8,85 \cdot 10^{-12} \cdot 1 \cdot 13,33 \cdot 10^5 = 1,18 \cdot 10^{-5} \text{ C/m}^2$$

$$D_2 = \epsilon_2 \cdot E_2 = \epsilon_0 \cdot \epsilon_{r2} \cdot E_2 = 8,85 \cdot 10^{-12} \cdot 4 \cdot 13,33 \cdot 10^5 = 4,72 \cdot 10^{-5} \text{ C/m}^2$$